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Chemistry 2nd Semester Final Exam Review

Chemistry . Second Semester Final Exam Review . Final Exam Learning Targets . Unit 5 • Perform 3-step grams-moles-moles-grams stoichiometry calculations. • Solve molarity problems . • Solve combined gas law problems. Unit 6 • Calculate the ΔH° for a chemical reaction from experiment data.

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Chemistry Second Semester Final Review

College Chemistry Second Semester Review Sheet Winter, Dr. Wicks Chapter 6: Thermochemistry • I understand the law of conservation of energy and I can give examples of various energy types. • I can distinguish between potential energy and kinetic energy and give examples for each.

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Chemistry Second Semester Review Answers

Chemistry 2nd Semester Final Exam Review Chemistry Second Semester Review Chpt 10 Questions 1) What is Avogadro's Number? 2) Draw a basic mole map for one chemical substance. 3) Calculate the molar mass of one compound of your choice. 4) Describe how to determine: mole->mass, mole->representative particle and mole->volume of your choice.

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2nd Semester Chemistry-575 Final Exam Review Multiple Choice Identify the choice that best completes the statement or answers the question. ____

1. How many moles of tungsten atoms are in 4.8×10^{25} atoms of tungsten? a. 8.0×10^2 moles c. 1.3×10^{-1} moles b. 8.0×10^1 moles d. 1.3×10^{-2} moles
____ 2.

2nd Semester Chemistry-575 Final

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Exam Review

CHEMISTRY REVIEW SHEET SECOND SEMESTER Chemistry Utica Community Schools - Semester Two Review

Directions: You will be given a periodic table with polyatomic ions and any formulas you may need. If you complete the entire review sheet by the day of your exam you will be eligible for the curve should one be applied.

CHEMISTRY REVIEW SHEET SECOND SEMESTER

Second Semester Chemistry Review. Chapter 9 Covalent Bonding. 1. Write the formulas for the following compounds. a. Calcium chloride b. Cobalt (II) carbonate c. Chromium (III) bromide . d. carbon tetrachloride e. barium nitrate f. ammonium hydroxide. 2. Write the names for each of these compounds. a. AlF_3 d. PCl_5 . b. Li_2CO_3 e. P_2O_5 . c. $Zn(NO_3)_2$...

Second Semester Chemistry Review
2nd Semester Chemistry-575 Final Exam

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Review Answer Section MULTIPLE CHOICE 1. ANS: B PTS: 1 DIF: L2 REF: p. 290 | p. 291 OBJ: 10.1.2 Relate Avogadro's number to a mole of a substance. 2. ANS: B Mole ratio is 2 moles Al for 3 moles FeO. Therefore 1 mole Al would only need 1.5 moles of FeO. No need for mole island if you have nice whole numbers.

2nd Semester Chemistry-575 Final Exam Review Answer Section

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